



(19) Europäisches Patentamt
European Patent Office
Office européen des brevets



(11) Publication number:

0 482 513 A2

(12)

EUROPEAN PATENT APPLICATION

(21) Application number: 91117731.9

(61) Int. Cl.⁵: **C09D 5/24, C09D 5/46,
B05D 1/04**

(22) Date of filing: **17.10.91**

(30) Priority: **22.10.90 US 600799**

(43) Date of publication of application:
29.04.92 Bulletin 92/18

(84) Designated Contracting States:
AT BE DE DK ES FR GB IT NL SE

(71) Applicant: **PPG INDUSTRIES, INC.**
One PPG Place
Pittsburgh Pennsylvania 15272(US)

(72) Inventor: **Hartman, Marvis Edgar**
144 Rawlins Run Road
Pittsburgh, Pennsylvania 15238(US)
Inventor: **Rei, Massimo P.**
Via Monterotondo 10
I-15100 Alessandria(IT)
Inventor: **Castagnone, Michela**
Via Monte San Michele 20
I-13039 Trino Vercellese(IT)
Inventor: **Hamay, Steven A.**
110 Eden Way Court
Evans City, Pennsylvania 16033(US)

(74) Representative: **Modiano, Guido et al**
MODIANO, JOSIF, PISANTY & STAUB
Baaderstrasse 3
W-8000 München 5(DE)

(54) **Light colored conductive coating and method and composition for application thereof.**

(57) A conductive primer coating composition comprising a resinous film forming binder, a fibrous, carbonaceous material and a light colored pigment. This coating composition may be applied to plastic or other substrates having low electrical conductivity to form a light colored, conductive coating. The application of this first coating and subsequent coatings by the electrostatic method of application is facilitated.

EP 0 482 513 A2

Background of the Invention

1. Field of the Invention

5 The present invention relates to coatings and coating compositions and more particularly to coating compositions for use on plastics and other low conductivity substrates and to methods for their application.

2. Brief Description of the Prior Art

10 The use of parts formed from sheet molding compound (SMC) and other plastics in automobile bodies presents a number of advantages such as light weight and resistance to corrosion. One problem with the use of such parts, however, is that they are more difficult than metal to electrostatically spray paint. This difficulty is at least in part attributable to the fact that static electricity which develops as a result of the use of electrostatic spraying equipment can not be easily dissipated through the nonconductive parts. In order
15 to solve this problem and impart conductivity to such parts it has been the practice to initially paint the parts with a primer which is rendered conductive by means of relatively large amounts of conductive carbon black in its composition. Such primers, however, have a resultant disadvantage in that they tend to be black or very dark gray in color. It may, therefore, be difficult to obtain good finishing coat hiding of the primer color.

20 Summary of the Invention

The coating composition of the present invention is a coating composition for use on plastic substrates which is made up of a resinous film forming binder, a quantity of fibrous, carbonaceous material and a white and/or other light colored pigment. The fibers in the fibrous, carbonaceous material are preferably intertwined. The fibrous, carbonaceous material also preferably comprises carbon fibrils. The carbon fibrils preferably are discrete, cylindrical particles which are characterized by a substantially constant diameter between about 3.5 and about 70 nanometers and a length about 100 times their diameters. Their length is greater than about 100 times the diameter. They also have an outer region of multiple essentially continuous layers of ordered carbon atoms and a distinct inner core region. Each of the layers and the core is disposed substantially concentrically about the cylindrical axis of the fibril, and the entire fibril is substantially free of a thermal carbon overcoat. These carbon fibrils are used in an amount from 0.05 to 5.00 percent by weight of the entire composition. The white and/or other light colored pigment is used in an amount of from 20 to 70 percent by weight of the composition. The coating composition will preferably be
30 electrostatically applied to a plastic or other low conductivity substrate. When this coating is used as a primer, it will be light colored and will have sufficient electrical conductivity to facilitate application of the topcoat by electrostatic spraying. In addition to being used as a primer, the coating composition of this invention can be advantageously used to apply a finishing coating over a primer or directly over a substrate.

40 Detailed Description

A preferred binder for use in the coating composition of the present invention comprises a saturated resin having pendent hydroxyl groups and an aminoplast curing agent. Examples of satisfactory resins are saturated polyesters, saturated polyurethanes and saturated polymers of ethylenically unsaturated monomers, all having hydroxyl functionality. Aromatic polymers, although being unsaturated in a strictly technical sense, fall within the term "saturated" as used herein because such saturation does not enter into free radical polymerization reactions.

Preferably this binder will be a resinous composition which comprises a polyol component containing at least 25 percent of a non-gelled polymeric polyol selected from the class, including mixtures thereof, consisting of polyester polyols, polyether polyols and polyurethane polyols, said non-gelled polymeric polyol having a hydroxyl value of less than 200 and a cured glass transition temperature less than 0°C.

One type of such resinous composition is further described in U.S. Patent No. 4,154,891, the contents of which are incorporated herein by reference.

Another suitable binder for use in the coating composition of the present invention would be a polyepoxide and a polyacid curing agent as is disclosed in U.S. Patent No. 4,737,403, the contents of which are incorporated herein by reference.

A light colored pigment is also be included in the coating composition of the present invention. Suitable light colored inorganic pigments would include titanium dioxide, zinc chromates, white lead, zinc oxide, zinc sulfide, antimony oxides, magnesium carbonate, calcium carbonate, calcium sulfate, barium sulfate, magnesium silicate, aluminum silicate, kaolinates, silicas and micas. It is believed that organic pigments, particularly the light colored yellows, could also be usefully employed in this composition.

Two or more light colored pigments may also be used together in the coating composition of the present invention. When two light colored pigments are used together in a primer, it may be particularly advantageous to use a white pigment and a pigment having a hue which is similar to the hue in the color of the finishing coat which will be applied over the primer coating of the present invention.

Typically, the total amount of white and/or other light colored pigment incorporated in the coating composition is in the amount of 20 to 70 percent by weight of the entire composition.

For the purpose of this description a light colored pigment would include a white colored pigment and would ordinarily be considered to have a color which has a Munsell value of 4.5 or more as is defined in the ISCC-NBS Method of Designating Color (Kelly 1955, 1976). Under that system, value is that quality of color which is described by the words light or dark and which relates the color to a gray of a similar lightness. Munsell value is a number from 0 to 10 with medium gray being from 4.5 to 6.5, light gray being from 6.5 to 8.5 and white being above 8.5. Dark gray and black would be below 4.5 to 2.0 and 2.0 to 0, respectively.

The fibers in the fibrous, carbonaceous material are preferably intertwined. The fibrous, carbonaceous material which is included in the coating composition of the present invention also preferably comprises carbon fibrils. For the purpose of this disclosure, carbon fibrils are carbon filaments with at least some graphitic domains which will normally be oriented with the c-axes of the graphite perpendicular to the fiber axis and which are prepared by contacting a metal catalyst with a carbon-containing gas. Examples of suitable carbon-containing gases would include ethylene, propylene, propane, methane, carbon monoxide, benzene, naphthalene, toluene and oxygenated hydrocarbons. A description of the crystal structure of graphite including its c-axis is included in Encyclopedia of Chemical Technology, Kirk-Othmer, 2nd Edition, Interscience, 1964, Vol. 4 pages 161 to 162, the contents of which are incorporated herein by reference. The carbon fibrils which are preferably used in the coating composition of the present invention are described in U.S. Patent No. 4,663,230 and International Patent Application published under the Patent Cooperation Treaty (PCT) No. WO 89/07163 (dated August 10, 1989), the contents of both of said references also being incorporated herein by reference. Essentially, such preferred carbon fibrils are characterized by a substantially constant diameter between about 3.5 and about 70 nanometers, e.g. between about 7 and 25 nanometers, length greater than about 100 times the diameter, an outer region of multiple essentially continuous layers of ordered carbon atoms and a distinct inner core region, each of the layers and core disposed substantially concentrically about the cylindrical axis of the fibril. Preferably, the entire fibril is substantially free of thermal carbon overcoat. The inner core of the fibril may be hollow or may contain carbon atoms which are less ordered than the ordered carbon atoms of the outer region, which are graphitic in nature.

When used as a primer, the coating composition may also contain optional ingredients such as plasticizers, flow control agents and other formulating additives. If these ingredients are present, they are usually present in amounts of up to 40 percent by weight based on total weight of the primer coating composition.

The present invention is particularly useful in coating compression molded and fiber-reinforced plastics and primarily the polyester variety which are known in the art as sheet molding compounds (SMC) or bulk molding compounds (BMC). Such materials are described in numerous publications and patents including the annually published Modern Plastics Encyclopedia, The Encyclopedia of Chemical Technology, Kirk-Othmer, 2nd edition, Vol. 20, pages 791-839, as well as U.S. Patent No. 3,184,527. It is to be appreciated, however, that the coating of the present invention seems to be useful with other compression and injection molded plastics both thermosetting and thermoplastic, including, epoxides, phenolics, silicones, aminoplasts, polyurethanes, polyethylene, polystyrene, polypropylene, thermoplastic acrylics, polyvinyl chloride and various polymers and various copolymers of acrylonitrile and polybutadiene. In particular, it is believed that the coating of the present invention will be useful on reaction injection molded (RIM) plastics and thermoplastic elastomers (TPEs) including thermoplastic polyolefins (TPOs), which materials are described in the aforesaid Modern Plastics Encyclopedia. Also, fibers other than glass fibers, for example, boron fibers, may be used to reinforce the above mentioned plastics. It is also believed that the coating of the present invention will be useful as a primer or as a finish coating on other low conductivity substrates consisting of materials such as wood, glass and other ceramics, paper, textiles, fiberboard, plasterboard and

particle board. For the purpose of this disclosure, such low conductivity substrates will, in general, be considered to have electrical resistance of more than 1,000 megaohms per inch (39,370 megaohms per meter) between surface points.

The coating composition can be applied to the substrate by any of the conventional coating techniques such as brushing, spraying, dipping or flowing, but it is preferred that spray application be used. Any of the known spray techniques may be employed such as compressed air spraying, electrostatic spraying, in either manual or automatic methods. If spraying is used, the viscosity of the primer coating composition should be adjusted with additional solvent if necessary such that it has a No. 4 Ford cup viscosity of 12-25 seconds, usually on the order of about 18 seconds at 75° F. (24° C.). The coating composition of the present invention is particularly advantageously applied by electrostatic spraying which is described, for example, in Encyclopedia of Chemical Technology, Kirk-Othmer, 3rd Edition, Wiley-Interscience, 1978, Vol. 6, pages 417 to 418 and The Encyclopedia of Polymer Science and Technology, Vol. 3, pages 797 to 801 and 825 to 826, the contents of both of said references being incorporated herein by reference. Essentially, in electrostatic spraying, particles of coating are electrically charged and the substrate to be coated is oppositely charged. The coating is sprayed toward the substrate, and because of this difference in charge the coating is attracted to the substrate. It will be understood that the coating composition of the present invention may be advantageously employed in electrostatically spraying a composition to apply either a primer coating directly over the substrate, a finishing coating directly over the primer coating or a single coating applied directly over the substrate. If sufficient fibrous, carbonaceous material is used so that the coating composition is, itself, electrically conductive, it is believed that the electrical charge applied to a coating composition being applied directly to a low conductivity substrate will tend to dissipate as soon as it is applied to the substrate and thereby reduce repulsion between the coating composition already applied to the substrate and charged particles of coating composition be sprayed toward the substrate.

After the coating composition has been applied to the substrate, the composition is heated to cure the coating. In the curing operation, the solvents are driven off, a substantially continuous coating is formed, and the coating is crosslinked. The heating operation is usually at low temperatures. Typically, temperatures from as low as 225° F. (107° C.) to as high as 325° F. (163° C.) can be used. However, the coating can be exposed to higher temperatures, as high as 400° F. (204° C.) without adverse effects. The time for curing is typically from about 15 to 60 minutes, more normally on the order of about 20 to 40 minutes. The thickness of the cured primer coating is typically from about 0.2 to 10 and preferably from about 0.5 to 2 mils.

After curing, the color of the coating should have a Munsell value of more than about 3.5 and preferably more than about 5.5. The coating should also have electrical resistance between two electrodes positioned at points on the surface of the coating of less than about 760 megaohms per inch (29,921.2 megaohms per meter) and preferably less than about 600 megaohms per inch (23,622 megaohms per meter) and more preferably less than about 190 megaohms per inch (7,480.3 megaohms per meter). The above mentioned three quantities of electrical resistance are equivalent, respectively, to readings of 105, 110 and 130 on the model no. 8333-00 sprayability meter available from the Ransburg Corporation of Indianapolis, Indiana.

After the primer has been applied to the plastic substrate and cured, a finishing or topcoat may be applied directly to the primer coat. While the finishing coat may advantageously be the coating composition of the present invention, it can also be any of the compositions known in the art for automotive applications or which is otherwise known for coating plastics. These compositions comprise a resinous binder and a pigment. Particularly useful resinous binders are acrylic polymers, polyesters, including alkyds and polyurethanes.

The pigments useful in the finishing coat include color pigments normally used in automotive applications such as titanium dioxide, iron oxide, chromium oxide, lead chromate and carbon black, and organic pigments such as phthalocyanine blue and phthalocyanine green. If desired, metallic pigments such as aluminum flake and metal oxide encapsulated micas can also be used in the finishing coat. Typically, the pigment is incorporated in the coating composition in amounts of about 0.5 to 25 percent by weight based on total weight of the finishing coat. If desired, the finishing coat may additionally contain other materials such as flow control agents, anti-oxidants, U.V. light absorbers and the like. These optional ingredients are typically present in amounts of up to 25 percent by weight based on total resin solids. It is found that the use of the above described primer coating of the present invention allows the primer to be effectively hidden by the finishing coat.

The finishing coat can be applied to the primer by conventional means such as brushing, spraying, flow coating and the like, but they are most often applied by spraying. The usual spray techniques and equipment for air spraying, electrostatic spraying in either manual or automatic methods can be used. It is found that the conductive primer coating of the present invention will facilitate the application of the topcoat with electrostatic spray equipment.

After application of the finishing coat to the basecoat, it is heated to coalesce and cure the coating. In the coating operation, solvents are driven off, a substantially continuous coating is formed and the coating is crosslinked. Typically, heating is from 180° to 325° F. (82° to 163° C.) for about 15 to 60 minutes depending upon the resinous vehicle and curing agent which are present in the finishing coat. Film thicknesses for the finishing coat are typically on the order of 1 to 5 mils.

The finishing coat can be applied as a one-coat system or alternately can be applied as a so-called color-clear system in which the pigmented or colored basecoat is first applied to the primer followed by the application of a clear topcoat over the colored basecoat. Such systems are being increasingly used in the automotive industry since they provide for excellent depth of color in the underlying basecoat as well as outstanding gloss, distinctness of image and durability in the coating.

The invention is illustrated by reference to the following examples wherein parts are parts by weight unless otherwise indicated.

Example A

A resin composition was made from the ingredients listed in Table 1 below.

Table 1

Charge A

<u>Ingredient</u>	<u>Parts by Weight</u>
Trimethylolpropane	197.7
Neopentyl glycol	1211.9
Water	177.8

Charge B

<u>Ingredient</u>	<u>Parts by Weight</u>
Adipic acid	181.7
Tetrahydrophthalic anhydride	1123.1
Dibutyltin dilaurate	2.7
Water	42.1

Charge C

<u>Ingredient</u>	<u>Parts by Weight</u>
Methylisobutyl ketone	200
Phosphoric acid (85% concentration)	107.5

Charge D

<u>Ingredient</u>	<u>Parts by Weight</u>
Aromatic naphthal ¹	650.3

¹ Available as SOLVESSO 100 from Exxon Corp.

Charge A was initially added to a reactor and melted. Charge B was then added and over a period of 6 hours was heated to 235° C. (455° F.) while the temperature at the top of the pack was held below 100° C. (212° F.). The reaction was terminated at an acid value of 3.0 and a Gardner-Holdt viscosity of X for an 80% solids solution in xylene. The product was cooled to 200° C. (392° F.) and vacuum was pulled to 3 mm./Hg to 160° C. (320° F.) for 15 minutes. The vacuum was broken with nitrogen. The methylisobutyl ketone was then added after which the rest of Charge C was added very slowly. The mixture was then heated to 160° C. (320° F.) and held at that temperature until an acid value of 26 was reached. All solvent was then

stripped and the product was cooled and Charge D was added. The completed product had a viscosity of Z₂-Z₃, a Gardner-Holdt color of 3 and an acid value of 29.3. After heating for 30 minutes at 150°C. (302°F.), it had a solids content of 74.3%.

5 Example B

A solvent blend was made by mixing the solvents shown in the following Table 2:

Table 2

10 <u>Ingredient</u>	<u>Parts by Weight</u>
SOLVESSO 100	71.00
Butyl acetate	15.00
15 Aromatic naphtha ²	7.00
Butanol	7.00
	100.00

20 ² Available as SOLVESSO 200 from Exxon Corp.

Example C

25 A resin was made from the ingredients listed in the following Table 3.

Table 3

30 <u>Ingredient</u>	<u>Parts by Weight</u>
Water	221.0
Neopentyl glycol	2043.0
Adipic acid	930.0
Isophthalic acid	792.0
35 Phthalic anhydride	235.0
Dibutyltin oxide	1595.0
SOLVESSO 100	539.0

40 All the ingredients except the SOLVESSO 100 were charged to a reactor and heated to 180°C. (356°F.) and held at that temperature for 30 minutes. The mixture was then heated to 200°C. (392°F.) and held at that temperature for 1 hour then heated to 220°C. (428°F.) and held at that temperature until an acid value of 2.9 was reached. The solvent was then added, and theoretical solids were 85%.

45 Example D

A white paste was made from mixing the ingredients listed in Table 4 by premixing under high speed, stirring for 30 minutes and then grinding in a sand mill for 40 minutes.

50

55

Table 4

<u>Ingredient</u>	<u>Parts by Weight</u>
Composition of Example A (74.2% solids)	375.00
Resin of Example C (80.0% solids)	150.00
Titanium dioxide	395.00
Silica	5.00
Silicone-free leveling agent ³	0.65
Solvent blend of Example B	<u>74.35</u>
	1000.00

³ Available as ADDITOL XL 480 from Hoechst Celanese Corp.

Example E

A black paste was made by mixing the ingredients listed in Table 5:

Table 5

<u>Ingredient</u>	<u>Parts by Weight</u>
Amine functional resin solution (35% by weight) ⁴	130.00
Carbon fibrils ⁵	11.00
Barium sulfate	313.00
Melamine-formaldehyde resin ⁶	330.00
SOLVESSO 100	<u>216.00</u>
	1000.00

⁴ Available as EFKA 47 from Efka Chemical BV of Hillegom, the Netherlands.

⁵ Available as BN-1000 graphite fibrils from Hyperion

Catalysis International, Inc. of Lexington, Massachusetts.

⁶ Available as CYMEL 303 from American Cyanamid Co.

Example F

A solvent blend was made from a mixture of the ingredients listed on the following Table 6:

Table 6

<u>Ingredient</u>	<u>Parts by Weight</u>
Xylene	40.00
Butyl acetate	25.00
Aromatic naphtha ⁷	<u>35.00</u>
	100.00

10 7 Available as SOLVESSO 200 from Exxon Corp.

15 Example 1

20 A light gray conductive primer was made by mixing the ingredients listed in the following Table 7.

Table 7

<u>Ingredient</u>	<u>Parts by Weight</u>
White paste of Example D	768.16
Black paste of Example E	<u>231.84</u>
	1000.00

25 The mixture was first thinned to No. 4 Ford Cup viscosity of 60 seconds with butyl acetate and subsequently thinned to a 30 seconds viscosity with the solvent solution of Example F. The resulting composition had the following characteristics:

Solid content (spray viscosity):	68-70%
Polyester ratio : 771368/771344:	70/30
Polyester/Melamine ratio:	80/20
Pigment/Binder ratio:	50/50
% carbon fibrils (by weight)	0.335

30 This composition was electrostatically sprayed onto a panel formed from sheet molding compound (SCM) to a thickness of 2 to 2.5 mils. After it was cured for 30 minutes at 140°C. (284°F.), the cured coating had the following characteristics:

35 Electrical resistance: <190 megaohms per inch (<7,480.3 megaohms per meter)
 Munsell value: 7
 Thickness 1.5 to 2 mils

40 Claims

- 45 1. A coating composition containing a resinous film forming binder, a fibrous, carbonaceous material and a light colored pigment.
- 50 2. The coating composition of claim 1 wherein the light colored pigment is of a color having a Munsell value of more than about 4.5.
- 55 3. The coating composition of claim 2 wherein the light colored pigment is of a color having a Munsell value of more than about 6.5.
- 60 4. The coating composition of claim 3 wherein the light colored pigment is white.

5. The coating composition of claim 1 wherein the light colored pigment is selected from the group consisting of titanium dioxide, zinc chromates, white lead, zinc oxide, zinc sulfide, antimony oxides, magnesium carbonate, calcium carbonate, calcium sulfate, barium sulfate, magnesium silicate, aluminum silicates, kaolinites, silicas and micas.
6. The coating composition of claim 1 wherein the fibrous, carbonaceous material comprises intertwined fibers.
7. The coating composition of claim 1 wherein the fibrous, carbonaceous material comprises carbon fibrils.
8. The coating composition of claim 7 wherein the carbon fibrils are essentially discrete and cylindrical and are characterized by a substantially constant diameter between about 3.5 and about 70 nanometers, length greater than about 100 times the diameter, an outer region of multiple essentially continuous layers of ordered carbon atoms and a distinct inner core region, each of the layers and core disposed substantially concentrically about the cylindrical axis of the fibril.
9. The coating composition of claim 8 wherein the cores of the carbon fibrils are hollow.
10. The coating composition of claim 8 wherein the inner core region of the fibrils comprises carbon atoms which are less ordered than the ordered carbon atoms of the outer region.
11. The coating composition of claim 8 wherein the ordered carbon atoms of the fibrils are graphitic.
12. The coating composition of claim 8 wherein the length of the fibrils is greater than about 100 times the diameter.
13. The coating composition of claim 8 wherein the length of the fibrils is between about 7 and 25 nanometers.
14. The coating composition of claim 8 wherein the inner core region of the fibrils has a diameter greater than about 2 nanometers.
15. The coating composition of claim 1 wherein the fibrous, carbonaceous material is present in an amount of from about 0.05 percent to about 5 percent by weight based on the weight of the entire primer composition.
16. The coating composition of claim 15 wherein the light colored pigment is present in the amount of from about 20 percent to about 70 percent by weight based on the weight of the entire composition.
17. The coating composition of claim 1 which is electrically conductive.
18. A conductive coating comprising a resinous film forming binder, a fibrous, carbonaceous material and a light colored pigment.
19. The conductive coating of claim 18 wherein the light colored pigment is of a color having a Munsell value of more than about 4.5.
20. The conductive coating of claim 18 wherein the coating contains light colored pigments having more than one color.
21. The conductive coating of claim 20 wherein the coating contains a white pigment and a pigment of a color having a hue and said hue is similar to a hue of a color of a coating overlaying said conductive coating.
22. The conductive coating of claim 18 wherein the fibrous, carbonaceous material comprises intertwined fibers.

23. The conductive coating of claim 18 wherein the fibrous, carbonaceous material comprises carbon fibrils.
24. The conductive coating of claim 18 wherein the carbon fibrils are essentially discrete and cylindrical and are characterized by a substantially constant diameter between about 3.5 and about 70 nanometers, length greater than about 100 times the diameter, an outer region of multiple essentially continuous layers of ordered carbon atoms and a distinct inner core region, each of the layers and core disposed substantially concentrically about the cylindrical axis of the fibril.
25. The conductive coating of claim 18 wherein the coating when cured is of a color having a Munsell value of more than about 3.5.
26. The conductive coating of claim 23 wherein the coating when cured is of a color having a Munsell value of more than about 5.5.
27. The conductive coating of claim 18 wherein the coating has electrical resistance between points on its surface of less than about 760 megaohms per inch (29,921.2 megaohms per meter).
28. The conductive coating of claim 27 wherein the coating has electrical resistance between points on its surface of less than about 600 megaohms per inch (23,622 megaohms per meter).
29. The conductive coating of claim 28 wherein the coating has electrical resistance between points on its surface of less than about 190 megaohms per inch (7,480.3 megaohms per meter).
30. The conductive coating of claim 18 wherein the coating is from about 0.2 mils to about 10 mils in thickness.
31. The conductive coating of claim 30 wherein the coating is from about 0.5 to about 2 mils in thickness.
32. The conductive coating of claim 18 wherein the coating is applied to a substrate having low electrical conductivity.
33. The conductive coating of claim 32 wherein the substrate is plastic.
34. The conductive coating of claim 33 wherein the substrate is formed from a sheet molding compound (SMC).
35. In a method for electrostatically applying a coating to a substrate by electrically charging particles of a coating composition with one charge and electrically charging the substrate to be coated with the opposite charge and delivering the particles of the coating composition to the substrate, wherein the improvement comprises using a coating composition which comprises a resinous film forming binder, a fibrous, carbonaceous material and a light colored pigment.
36. The method of claim 35 wherein the light colored pigment is of a color having a Munsell value of more than about 4.5.
37. The method of claim 35 wherein the fibrous, carbonaceous material comprises intertwined fibers.
38. The method of claim 36 wherein the fibrous, carbonaceous material comprises carbon fibrils.
39. The method of claim 38 wherein the carbon fibrils are essentially discrete and cylindrical and are characterized by a substantially constant diameter between about 3.5 and about 70 nanometers, length greater than about 100 times the diameter, an outer region of multiple essentially continuous layers of ordered carbon atoms and a distinct inner core region, each of the layers and core disposed substantially concentrically about the cylindrical axis of the fibril.
40. The method of claim 35 wherein the substrate to which the coating composition is applied has low electrical conductivity.

41. The method of claim 40 wherein the substrate to which the coating is applied is plastic.
42. The method of claim 41 wherein the substrate to which the coating is applied is formed from a sheet molding compound (SMC).
- 5 43. The method of claim 35 wherein the coating composition is applied directly over the substrate as a primer coat.
- 10 44. The method of claim 43 wherein the coating composition is applied directly over the primer coat as a finishing coat.
45. The method of claim 40 wherein the coating composition is applied directly over the substrate as a single coating.

15

20

25

30

35

40

45

50

55

THIS PAGE BLANK (USPTO)



(19) Europäisches Patentamt
European Patent Office
Office européen des brevets



(11) Publication number:

0 482 513 A3

(12)

EUROPEAN PATENT APPLICATION

(21) Application number: 91117731.9

(51) Int.Cl.: C09D 5/24, C09D 5/46,
B05D 1/04

(22) Date of filing: 17.10.91

(30) Priority: 22.10.90 US 600799

(71) Applicant: PPG INDUSTRIES, INC.
One PPG Place
Pittsburgh Pennsylvania 15272(US)

(43) Date of publication of application:
29.04.92 Bulletin 92/18

(72) Inventor: Hartman, Marvis Edgar
144 Rawlins Run Road
Pittsburgh, Pennsylvania 15238(US)
Inventor: Rei, Massimo P.
Via Monterotondo 10
I-15100 Alessandria(IT)
Inventor: Castagnone, Michela
Via Monte San Michele 20
I-13039 Trino Vercellese(IT)
Inventor: Hamay, Steven A.
110 Eden Way Court
Evans City, Pennsylvania 16033(US)

(84) Designated Contracting States:
AT BE DE DK ES FR GB IT NL SE

(74) Representative: Modiano, Guido et al
Modiano & Associati S.r.l. Baaderstrasse 3
W-8000 München 5(DE)

(88) Date of deferred publication of the search report:
13.01.93 Bulletin 93/02

(54) Light colored conductive coating and method and composition for application thereof.

(57) A conductive primer coating composition comprising a resinous film forming binder, a fibrous, carbonaceous material and a light colored pigment. This coating composition may be applied to plastic or other substrates having low electrical conductivity to form a light colored, conductive coating. The application of this first coating and subsequent coatings by the electrostatic method of application is facilitated.



European Patent
Office

EUROPEAN SEARCH REPORT

Application Number

DOCUMENTS CONSIDERED TO BE RELEVANT

EP 91117731.9

Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. CL.S)
X	<u>DE - A - 3 420 968</u> (KERAMCHEMIE GMBH) * Claims; example * --	1, 4, 5	C 09 D 5/24 C 09 D 5/46 B 05 D 1/04
X	<u>EP - A - 0 079 589</u> (DAIKIN KOGYO CO., LTD.) * Claims; page 11, para- graph 3 * ----	1, 4, 5	
TECHNICAL FIELDS SEARCHED (Int. CL.S)			
C 09 D 5/00 B 05 D 1/00			

The present search report has been drawn up for all claims

Place of search	Date of completion of the search	Examiner
VIENNA	30-10-1992	PAMMINGER
CATEGORY OF CITED DOCUMENTS		
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document